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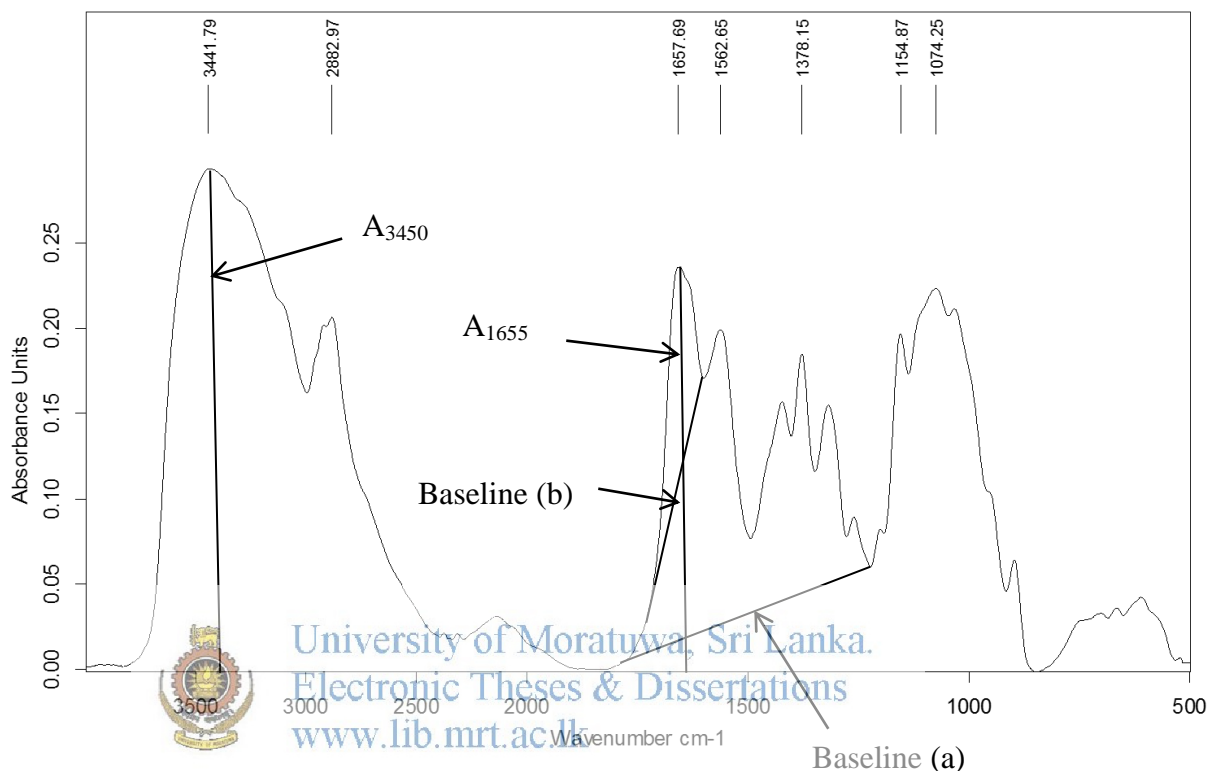
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**[Appendix - I: Determination of Degree of deacetylation]**



Baseline (a) equation

$$DD = 100 - \left[ \left( \frac{A_{1655}}{A_{3450}} \right) \times \frac{100}{1.33} \right]$$

Baseline (b) equation

$$DD = 100 - \left[ \left( \frac{A_{1655}}{A_{3450}} \right) \times 115 \right]$$

The acid base titration method was used to determine the DD from the amino group content in chitosan. Dry chitosan (0.3g) was dissolved in 30ml of HCL standard solution (0.1M). Methyl orange and aniline blue mixing indicators were added. A standard solution of NaOH (0.1M) was used for titration until the solution became blue green. The following formulas were used to calculate the DD of the product.

$$(-\text{NH}_2)\% = \frac{0.016(C_1V_1 - C_2V_2)}{W} \times 100$$

$$\text{DD}\% = \frac{203(-\text{NH}_2\%)}{16 + 42(-\text{NH}_2\%)} \times 100$$

Where  $C_1$ ,  $V_1$ ,  $C_2$ , and  $V_2$  are the concentrations and volumes for the HCl standard solution and NaOH standard solution, respectively, and  $W$  is the weight of the sample.



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Sample	FTIR base line a method	FTIR base line b method	Titration method
Low DD	72	88.6	94.2
High DD	85.6	93.6	98.5

***[Appendix - II: FTIR Characterization of heavy metal adsorbed chitosan]***

Characteristic peaks which affect the adsorption of heavy metals are as follows,

- N-H stretching band of amine group – 3270- 3300  $\text{cm}^{-1}$
- C=O stretching of amide I – 1655  $\text{cm}^{-1}$
- C-N stretching of amide I – 1625  $\text{cm}^{-1}$
- N-H bending band due to the presence of  $\text{NH}_2$  band – 1590  $\text{cm}^{-1}$

Due to the presence of number of peaks in the near wave number values, it is very difficult to find the actual intensity of the particular peak. For that, FTIR graphs were normalized relative to the O-H Stretching peak (Because OH doesn't involve with the adsorption process) and then the graph was deconvoluted to find the correct intensity.

When heavy metal adsorbed in to  $\text{NH}_2$ , shift and broadening of the peak can be observed. That can be analyzed by calculating the full width at half maximum of  $\text{NH}_2$  absorption band as represent in Figure 1.



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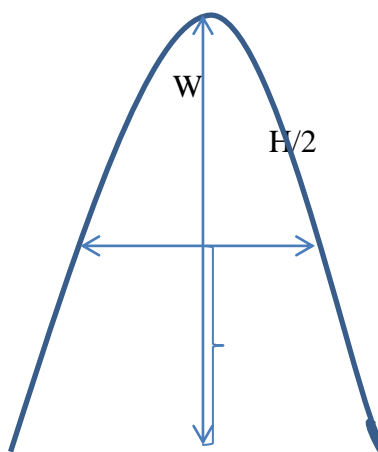


Figure 1. Representation of full width at half

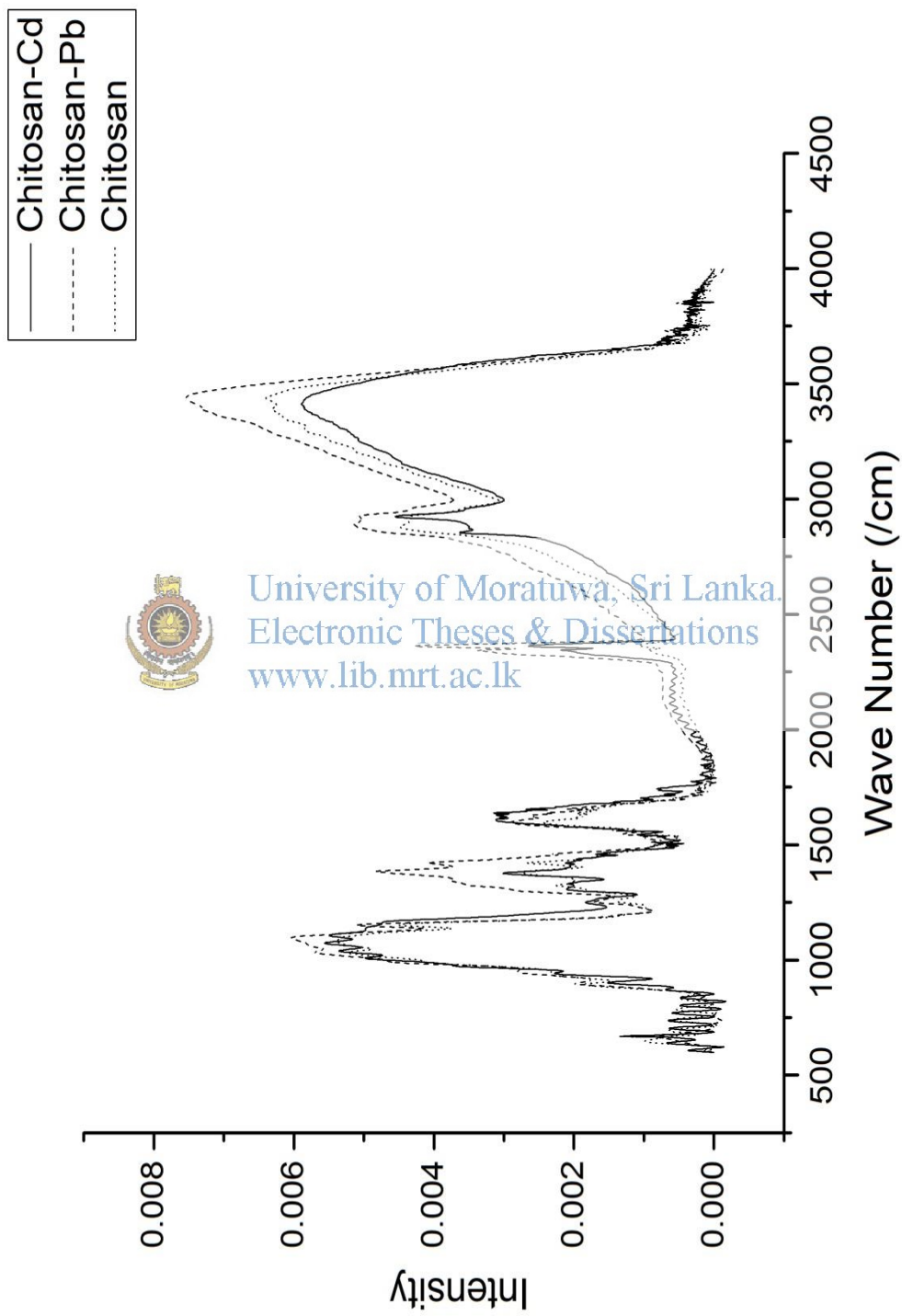


Figure 2. FTIR spectrum comparison of different chitosan samples

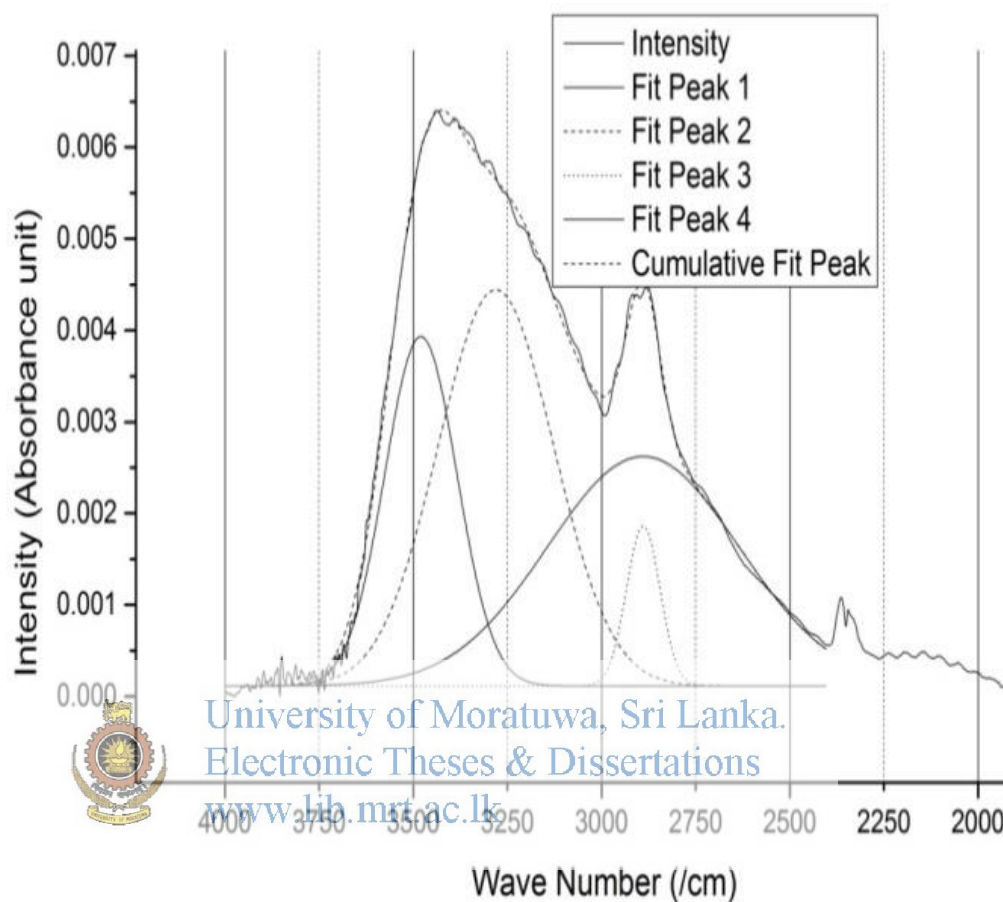


Figure 3. Deconvolute graph of chitosan

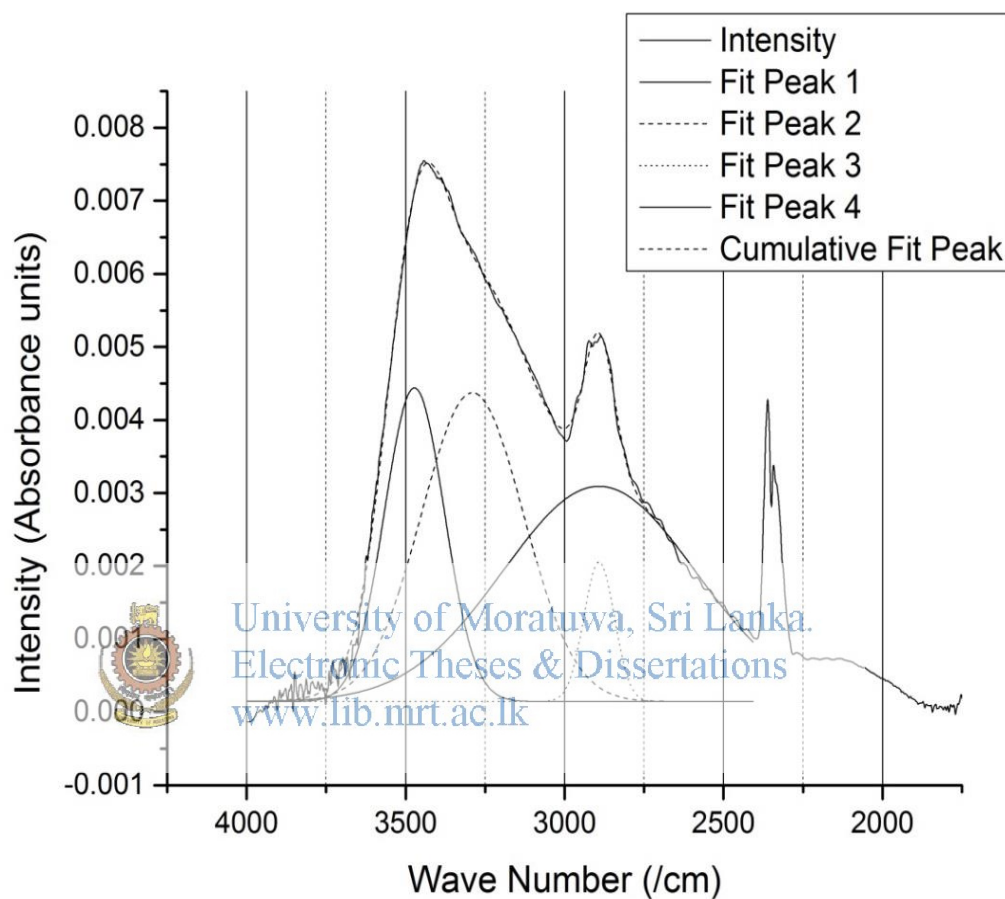


Figure 4. Deconvolute graph of Cd-chitosan

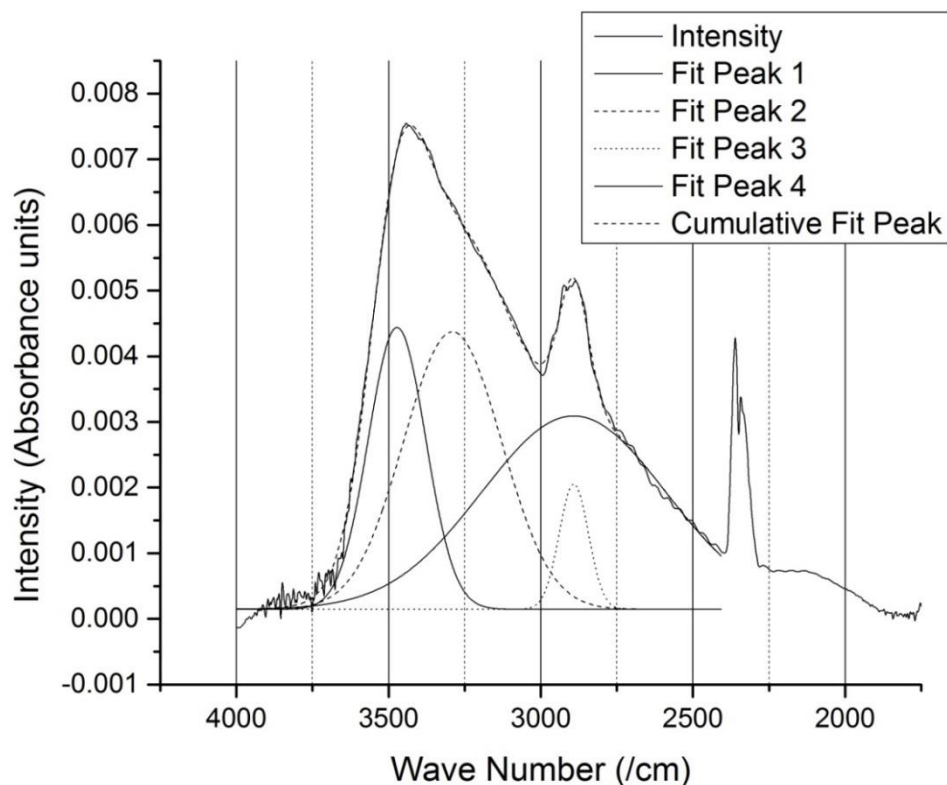


Figure 5. Deconvolute graph of Pb-chitosan

In all above three graphs, peak 2 is responsible for stretching of N-H bonds. So, if the Cd and Pb were connected to the NH<sub>2</sub> groups by forming complex structure, broadening of the peak can be observed. In the following Table 1, full widths at half maximum (W) data are given.

Table 1. Full width at half maximum values for chitosan samples

	W (nm)
Pure Chitosan	360.78
Cd Adsorbed Chitosan	377.97
Pb Adsorbed Chitosan	378.97

Finally, it can be concluded that the cadmium and lead were adsorbed by the NH<sub>2</sub> groups in chitosan.

[Appendix - III: Adsorption test results]

Cd

	Time (min)	Final concentration of solution (ppm)	Adsorbed amount into chitosan (ppm)	Adsorption capacity (mg/g)
<b>Low DD</b> pH- 5.5 28±2 <sup>0</sup> C 0.1 g dose 50ppm	5	41.8	8.2	4.1
	10	41	9	4.5
	15	38.333	11.667	5.8335
	30	34.433	15.567	7.7835
	45	34.425	15.575	7.7875
	60	34.467	15.533	7.7665
	90	33.966	16.034	8.017
	120	33.733	16.267	8.1335
	150	33.733	16.267	8.1335
	180	33.766	16.234	8.117
	240	30.425	19.575	9.7875
300	30.426	19.574	9.787	
<b>High DD</b> pH- 5.5 28±2 <sup>0</sup> C 0.1 g dose 50ppm	5	34.925	15.075	7.5375
	10	34.45	15.55	7.775
	15	31.266	18.734	9.367
	30	29.867	20.133	10.0665
	45	29.4	20.6	10.3
	60	29.433	20.567	10.2835
	90	29.233	20.767	10.3835
	120	28.867	21.133	10.5665
	150	27.433	22.567	11.2835
	180	27.467	22.533	11.2665
	240	26.874	23.126	11.563
300	26.548	23.452	12.226	
<b>High pH-6.5</b> 28±2 <sup>0</sup> C 0.1 g dose 50ppm	5	39.15	10.85	5.425
	10	37.625	12.375	6.1875
	15	32.533	17.467	8.7335
	30	28.123	21.877	10.9385
	45	26.98	23.02	11.51
	60	26.15	23.85	11.925
	90	25.456	24.544	12.272
	120	24.98	25.02	12.51
	150	24.625	25.375	12.6875
	180	24.533	25.467	12.7335
	240	24.523	25.477	12.7385
300	24.321	25.679	12.8395	



<b>Low pH-3.5 28±2<sup>0</sup>C 0.1 g dose 50ppm</b>	5	45.725	4.275	2.1375
	10	44.625	5.375	2.6875
	15	44.499	5.501	2.7505
	30	44.267	5.733	2.8665
	45	43.733	6.267	3.1335
	60	43.267	6.733	3.3665
	90	42.2	7.8	3.9
	120	41.266	8.734	4.367
	150	40.267	9.733	4.8665
	180	40.266	9.734	4.867
	240	40.261	9.739	4.8695
300	40.256	9.744	4.872	
<b>106 microns pH- 6.5 28±2<sup>0</sup>C 0.1 g dose of chitosan 50ppm</b>	5	38	12	6
	10	35.64	14.36	7.18
	15	35.36	14.64	7.32
	30	32.64	17.36	8.68
	45	31.42	18.58	9.29
	60	29.26	20.74	10.37
	90	28.4	21.6	10.8
	120	26.06	23.94	11.97
	150	24.86	25.14	12.57
	180	22.36	27.64	13.82
	240	22.3	27.7	13.85
300	22.06	27.94	13.97	
<b>75 microns pH- 6.5 28±2<sup>0</sup>C 0.1 g dose of chitosan 50ppm</b>	5	35.8	14.2	7.1
	10	35.32	14.68	7.34
	15	34.12	15.88	7.94
	30	32.02	17.98	8.99
	45	31.58	18.42	9.21
	60	28.16	21.84	10.92
	90	27.36	22.64	11.32
	120	25.02	24.98	12.49
	150	23.06	26.94	13.47
	180	21.06	28.94	14.47
	240	21.16	28.84	14.42
300	20.8	29.2	14.6	

<b>25 ppm</b> pH- 6.5 28±2 <sup>0</sup> C 0.1 g dose of chitosan	5	19.19	5.81	2.905
	10	19.4	5.6	2.8
	15	17.18	7.82	3.91
	30	11.41	13.59	6.795
	45	10.35	14.65	7.325
	60	10.81	14.19	7.095
	90	10.32	14.68	7.34
	120	9.3	15.7	7.85
	150	8.5	16.5	8.25
	180	7.85	17.15	8.575
	240	7.64	17.36	8.68
	300	6.5	18.5	9.25
<b>5 ppm</b> pH- 6.5 28±2 <sup>0</sup> C 0.1 g dose of chitosan	5	0.806	4.194	2.097
	10	0.798	4.202	2.101
	15	0.66	4.34	2.17
	30	0.798	4.202	2.101
	45	0.706	4.294	2.147
	60	0.652	4.348	2.174
	90	0.672	4.328	2.164
	120	0.602	4.398	2.199
	150	0.638	4.362	2.181
	180	0.592	4.408	2.204
	240	0.576	4.424	2.212
	300	0.562	4.438	2.219
<b>0.05 g</b> pH- 6.5 28±2 <sup>0</sup> C 50ppm	5	42.65	7.35	7.35
	10	40.875	9.125	9.125
	15	41.725	8.275	8.275
	30	41.4	8.6	8.6
	45	37.95	12.05	12.05
	60	39.375	10.625	10.625
	90	38.7	11.3	11.3
	120	37.625	12.375	12.375
	150	36.25	13.75	13.75
	180	35.85	14.15	14.15
	240	33.94	16.06	16.06
	300	33.175	16.825	16.825

<b>0.025g</b> pH- 6.5 28±2 <sup>0</sup> C 50ppm	5	43.9	6.1	12.2
	10	43.725	6.275	12.55
	15	42.725	7.275	14.55
	30	42.6	7.4	14.8
	45	42.625	7.375	14.75
	60	42.075	7.925	15.85
	90	40.98	9.02	18.04
	120	40.175	9.825	19.65
	150	40.7	9.3	18.6
	180	39.35	10.65	21.3
	240	39.15	10.85	21.7
	300	38.95	11.05	22.1



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**Pb**

	Time (min)	Final concentration of solution (ppm)	Adsorbed amount into chitosan(ppm)	Adsorption capacity(mg/g)
<b>Low DD</b> pH- 3.5 28±2 <sup>0</sup> C 0.1 g dose of chitosan 50 ppm	5	27.665	22.335	11.1675
	10	26.795	23.205	11.6025
	15	27.325	22.675	11.3375
	30	26.955	23.045	11.5225
	45	24.605	25.395	12.6975
	60	22.65	27.35	13.675
	90	21.38	28.62	14.31
	120	17.265	32.735	16.3675
	150	20.525	29.475	14.7375
	180	16.48	33.52	16.76
	240	17.55	32.45	16.225
300	17.45	32.55	16.275	
<b>High DD</b> pH- 3.5 28±2 <sup>0</sup> C 0.1 g dose of chitosan 50 ppm	5	25.135	24.865	12.4325
	10	25.14	24.86	12.43
	15	25.37	24.63	12.315
	30	23.055	26.945	13.4725
	45	23.075	26.925	13.4625
	60	20.715	29.285	14.6425
	90	17.65	32.35	16.175
	120	14.526	35.474	17.737
	150	17.95	32.05	16.025
	180	12.42	37.58	18.79
	240	12.96	37.04	18.52
300	12.76	37.24	18.52	
<b>High pH-4.5</b> 28±2 <sup>0</sup> C 0.1 g dose of chitosan 50 ppm	5	25.135	24.865	12.4325
	10	25.14	24.86	12.43
	15	25.37	24.63	12.315
	30	23.055	26.945	13.4725
	45	23.075	26.925	13.4625
	60	20.715	29.285	14.6425
	90	17.65	32.35	16.175
	120	14.526	35.474	17.737
	150	17.95	32.05	16.025
	180	12.42	37.58	18.79
	240	12.96	37.04	18.52
300	12.76	37.24	18.52	

<b>Low pH-2.0</b> 28±2 <sup>0</sup> C 0.1 g dose of chitosan 50 ppm	5	34.805	15.195	7.5975
	10	32.57	17.43	8.715
	15	31.595	18.405	9.2025
	30	33.935	16.065	8.0325
	45	30.455	19.545	9.7725
	60	28.615	21.385	10.6925
	90	26.205	23.795	11.8975
	120	22.905	27.095	13.5475
	150	24.895	25.105	12.5525
	180	22.815	27.185	13.5925
	240	21.8	28.2	14.1
	300	21.68	28.32	14.16
<b>106 microns</b> pH- 4.5 28±2 <sup>0</sup> C 0.1 g dose of chitosan 50 ppm	5	24.5	25.5	12.75
	10	23.676	26.324	13.162
	15	23.652	26.348	13.174
	30	22.724	27.276	13.638
	45	22.752	27.248	13.624
	60	21.608	28.392	14.196
	90	19.0925	30.9075	15.45375
	120	18.9142	31.0858	15.5429
	150	17.6168	32.3832	16.1916
	180	16.593	33.407	16.7035
	240	16.9824	33.0176	16.5088
	300	15.732	34.268	17.134
<b>75 microns</b> pH- 4.5 28±2 <sup>0</sup> C 0.1 g dose of chitosan 50 ppm	5	24.98	25.02	12.51
	10	24.4	25.6	12.8
	15	23.2167	26.7833	13.39165
	30	22.4466	27.5534	13.7767
	45	21.349	28.651	14.3255
	60	19.621	30.379	15.1895
	90	18.315	31.685	15.8425
	120	17.7111	32.2889	16.14445
	150	17.5329	32.4671	16.23355
	180	16.869	33.131	16.5655
	240	15.732	34.268	17.134
	300	15.032	34.968	17.484



<b>25 ppm</b> pH- 4.5 28±2 <sup>0</sup> C 0.1 g dose of chitosan	5	6.04	18.96	9.48
	10	5.326	19.674	9.837
	15	5.136	19.864	9.932
	30	4.864	20.136	10.068
	45	4.624	20.376	10.188
	60	4.806	20.194	10.097
	90	4.214	20.786	10.393
	120	3.56	21.44	10.72
	150	3.97	21.03	10.515
	180	3.19	21.81	10.905
	240	2.282	22.718	11.359
	300	2.012	22.988	11.494
<b>5 ppm</b> pH- 4.5 28±2 <sup>0</sup> C 0.1 g dose of chitosan	5	0.775	4.225	2.1125
	10	0.551	4.449	2.2245
	15	0.513	4.487	2.2435
	30	0.54	4.46	2.23
	45	0.346	4.654	2.327
	60	0.318	4.682	2.341
	90	0.271	4.729	2.3645
	120	0.119	4.881	2.4405
	150	0.062	4.938	2.469
	180	0.008	4.992	2.496
	240	0.008	4.992	2.496
	300	0.007	4.993	2.4965



### **[Appendix – IV: Publications]**

1. IEEE Conference Publications

Unagolla, J. M., Adikary, S. U., “Adsorption of cadmium and lead heavy metals by chitosan biopolymer: A study on equilibrium isotherms and kinetics”, *Moratuwa Engineering Research Conference (MERCon), 2015*, pp 234-239, 2015, DOI: 10.1109/MERCon.2015.7112351

2. Unagolla J. M., Adikary S. U.; “Adsorption Characteristics of Cadmium and Lead heavy metal into Locally Synthesized Chitosan Biopolymer” *Tropical Agricultural Research Journal*, Vol 26(2), pp395-401,2014; ISSN: 1016.1422.

Available online at, [http://www.pgia.ac.lk/files/Annual\\_congress/ journal/v26](http://www.pgia.ac.lk/files/Annual_congress/ journal/v26)

3. Unagolla J.M., Adikary S. U.;“Adsorption of Lead heavy metal ions by Chitosan Biopolymer: Kinetics and Equilibrium” *108<sup>th</sup> Annual transactions of Institute of Engineers Sri Lanka*, Vol 1 – Part B, pp 155-162, 2014.

4. Unagolla, J. M., Adikary, S. U.; “Study of Adsorption Characteristics of Cadmium into Chitosan Biopolymer to be used for Waste Water Treatments” *107<sup>th</sup> Annual transactions of Institute of Engineers Sri Lanka*, Vol 1- Part B, pp 313-319, 2013.

Available online at, <http://www.scribd.com/doc/202126465/IESL-Technical-Papers-Oct-2013#scribd>